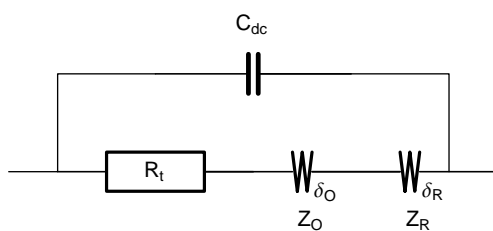


# Electrochemical Impedance Spectroscopy Library



ER@SE

December 28, 2003



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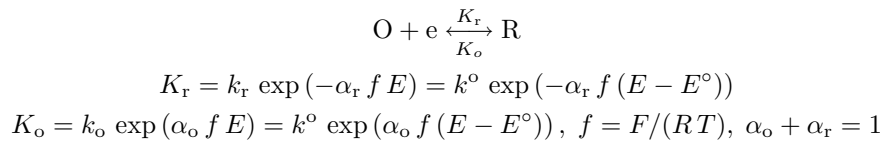
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# Chapter 1

## Reactions involving soluble species only

### 1.1 Redox reaction (E)

#### 1.1.1 Mechanism



#### 1.1.2 Kinetic equations

##### Transformation rates

$$v_{\text{O}}(t) = -v(t), v_{\text{R}}(t) = v(t)$$

##### Mass balance equations

$$\text{Flux of soluble species : } J_{\text{O}}(0, t) = v_{\text{O}}(t), J_{\text{R}}(0, t) = v_{\text{R}}(t)$$

##### Current density vs. reaction rate

$$i_{\text{f}}(t) = -F v(t)$$

##### Reaction rate

$$v(t) = -R(0, t) K_{\text{o}}(t) + O(0, t) K_{\text{r}}(t)$$

#### 1.1.3 Steady-state conditions

##### Steady-state equations

Soluble species

$$J_{\text{O}}(0) = -(O^* - O(0)) m_{\text{O}}, J_{\text{R}}(0) = -(R^* - R(0)) m_{\text{R}}$$

## Steady-state solutions

Soluble species

$$R(0) = \frac{R^* + K_r (R^*/m_O + O^*/m_R)}{1 + K_o/m_R + K_r/m_O}, \quad O(0) = \frac{O^* + K_o (R^*/m_O + O^*/m_R)}{1 + K_o/m_R + K_r/m_O}$$

Current density

$$i_f = \frac{F (K_o R^* - K_r O^*)}{1 + K_o/m_R + K_r/m_O}$$

### 1.1.4 Faradaic impedance

### 1.1.5 RDE (diffusion-convection)

#### Faradaic impedance

Faradaic impedance

$$Z_f(s) = R_{ct} + Z_O(s) + Z_R(s)$$

$$Z_f(s) = \frac{1 + K_r M_O(s) + K_o M_R(s)}{f F (R(0) K_o \alpha_o + O(0) K_r \alpha_r)}$$

Charge transfer resistance

$$R_{ct} = \frac{1}{f F (R(0) K_o \alpha_o + O(0) K_r \alpha_r)}$$

Concentration impedances (with  $\partial_X v = \frac{\partial v}{\partial X}$ )

$$Z_O = -\frac{\partial_O v M_O(s)}{\partial_E v} = K_r R_{ct} M_O(s)$$

$$Z_R = \frac{\partial_R v M_R(s)}{\partial_E v} = K_o R_{ct} M_R(s)$$

$$M_O(s) = \frac{1}{m_O} \frac{\text{th} \sqrt{\tau_{dO} s}}{\sqrt{\tau_{dO} s}}, \quad M_R(s) = \frac{1}{m_R} \frac{\text{th} \sqrt{\tau_{dR} s}}{\sqrt{\tau_{dR} s}}$$

Equivalent circuit (Fig. 1.1)

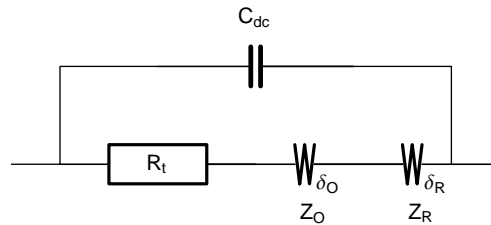


Figure 1.1: Equivalent circuit for the impedance of redox reactions (RDE).

### 1.1.6 Warburg conditions (semi-infinite linear diffusion)

Faradaic impedance

$$Z_f(s) = R_{ct} + Z_O(s) + Z_R(s)$$

only at the equilibrium potential:

$$E = E_{eq} = E^\circ + \frac{1}{f} \ln \frac{O^*}{R^*}$$

$$Z_f(s) = \frac{1 + K_r M_O(s) + K_o M_R(s)}{f F (R^* K_o \alpha_o + O^* K_r \alpha_r)}$$

Charge transfer resistance

$$R_{ct} = \frac{1}{f F (R^* K_o \alpha_o + O^* K_r \alpha_r)}$$

Concentration impedances

$$Z_O(s) = K_r R_{ct} M_O(s), Z_R(s) = K_o R_{ct} M_R(s)$$

$$M_O(s) = \frac{1}{\sqrt{s D_O}}, M_R(s) = \frac{1}{\sqrt{s D_R}}$$

Equivalent circuit (Fig. 1.2)

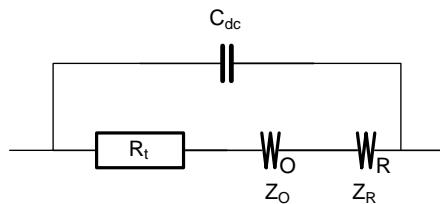
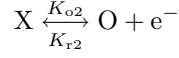
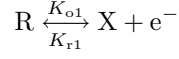


Figure 1.2: Equivalent circuit for the impedance of redox reactions: Warburg conditions.

## 1.2 EE reaction

### 1.2.1 Mechanism



$$K_{o1} = k_{o1} \exp(\alpha_{o1} f E) = k_1^o \exp(\alpha_{o1} f (E - E_1^o))$$

$$K_{r1} = k_{r1} \exp(-\alpha_{r1} f E) = k_1^o \exp(-\alpha_{r1} f (E - E_1^o)), \alpha_{o1} + \alpha_{r1} = 1$$

$$K_{o2} = k_{o2} \exp(\alpha_{o2} f E) = k_2^o \exp(\alpha_{o2} f (E - E_2^o))$$

$$K_{r2} = k_{r2} \exp(-\alpha_{r2} f E) = k_2^o \exp(-\alpha_{r2} f (E - E_2^o)), \alpha_{o2} + \alpha_{r2} = 1$$

### 1.2.2 Kinetic equations, without coupled homogeneous reactions

#### Transformation rates

$$v_R(t) = -v_1(t), v_X(t) = v_1(t) - v_2(t), v_O(t) = v_2(t)$$

#### Mass balance equations

Flux of soluble species

$$J_R(0, t) = v_R(t), J_X(0, t) = v_X(t), J_O(0, t) = v_O(t)$$

#### Current density vs. step rates

$$i_f(t) = F (v_1(t) + v_2(t))$$

#### Step rates

$$v_1(t) = R(0, t) K_{o1}(t) - X(0, t) K_{r1}(t), v_2(t) = X(0, t) K_{o2}(t) - O(0, t) K_{r2}(t)$$

### 1.2.3 Steady-state conditions

#### Steady-state equations

$$J_R(0) = -(R^* - R(0)) m_R, J_X(0) = -(X^* - X(0)) m_X, J_O(0) = -(O^* - O(0)) m_O$$

#### Steady-state solutions

Soluble species

$$R(0) = \left( R^* + \frac{R^* K_{r2}}{m_O} + \frac{X^* K_{r1}}{m_R} + \frac{X^* K_{r1} K_{r2}}{m_O m_R} + \frac{R^* K_{o2}}{m_X} + \frac{R^* K_{r1}}{m_X} + \frac{R^* K_{r1} K_{r2}}{m_O m_X} + \frac{O^* K_{r1} K_{r2}}{m_R m_X} \right) / \left( 1 + \frac{K_{r2}}{m_O} + \frac{K_{o1}}{m_R} + \frac{K_{o1} K_{r2}}{m_O m_R} + \frac{K_{o2}}{m_X} + \frac{K_{r1}}{m_X} + \frac{K_{r1} K_{r2}}{m_O m_X} + \frac{K_{o1} K_{o2}}{m_R m_X} \right)$$



$$X(0) = \left( X^* + \frac{X^* K_{r2}}{m_O} + \frac{X^* K_{o1}}{m_R} + \frac{X^* K_{o1} K_{r2}}{m_O m_R} + \frac{R^* K_{o1}}{m_X} + \frac{O^* K_{r2}}{m_X} + \frac{R^* K_{o1} K_{r2}}{m_O m_X} + \frac{O^* K_{o1} K_{r2}}{m_R m_X} \right) / \left( 1 + \frac{K_{r2}}{m_O} + \frac{K_{o1}}{m_R} + \frac{K_{o1} K_{r2}}{m_O m_R} + \frac{K_{o2}}{m_X} + \frac{K_{r1}}{m_X} + \frac{K_{r1} K_{r2}}{m_O m_X} + \frac{K_{o1} K_{o2}}{m_R m_X} \right)$$

$$O(0) = \left( O^* + \frac{X^* K_{o2}}{m_O} + \frac{O^* K_{o1}}{m_R} + \frac{X^* K_{o1} K_{o2}}{m_O m_R} + \frac{O^* K_{o2}}{m_X} + \frac{O^* K_{r1}}{m_X} + \frac{R^* K_{o1} K_{o2}}{m_O m_X} + \frac{O^* K_{o1} K_{o2}}{m_R m_X} \right) / \left( 1 + \frac{K_{r2}}{m_O} + \frac{K_{o1}}{m_R} + \frac{K_{o1} K_{r2}}{m_O m_R} + \frac{K_{o2}}{m_X} + \frac{K_{r1}}{m_X} + \frac{K_{r1} K_{r2}}{m_O m_X} + \frac{K_{o1} K_{o2}}{m_R m_X} \right)$$

Current density

$$i_f = \left( K_{o1} R^* + K_{o2} X^* + \frac{K_{o1} K_{o2} X^*}{m_R} + \frac{K_{o1} K_{r2} R^*}{m_O} + \frac{2 K_{o1} K_{o2} R^*}{m_X} - K_{r1} X^* - K_{r2} O^* - \frac{K_{r1} K_{r2} X^*}{m_O} - \frac{K_{o1} K_{r2} O^*}{m_R} - \frac{2 K_{r1} K_{r2} O^*}{m_X} \right) / \left( 1 + \frac{K_{r2}}{m_O} + \frac{K_{o1}}{m_R} + \frac{K_{o1} K_{r2}}{m_O m_R} + \frac{K_{o2}}{m_X} + \frac{K_{r1}}{m_X} + \frac{K_{r1} K_{r2}}{m_O m_X} + \frac{K_{o1} K_{o2}}{m_R m_X} \right)$$

## 1.2.4 Faradaic impedance

Faradaic impedance

$$Z_f(s) = R_{ct} + Z_O(s) + Z_R(s) + Z_X(s)$$

$$Z_f(s) = (1 + K_{o1} M_O(s)) (1 + K_{r2} M_R(s)) + (K_{o2} (1 + K_{o1} M_O(s)) + K_{r1} (1 + K_{r2} M_R(s))) M_X(s) / (f F (X(0) K_{r1} \alpha_{r1} (1 + K_{r2} M_R(s)) + 2 K_{o2} M_X(s)) + X(0) K_{o2} \alpha_{o2} (1 + 2 K_{r1} M_X(s)) + R(0) K_{r2} \alpha_{r2} (1 + 2 K_{r1} M_X(s)) + K_{o1} ((X(0) K_{o2} \alpha_{o2} + R(0) K_{r2} \alpha_{r2}) M_O(s) + O(0) \alpha_{o1} (1 + K_{r2} M_R(s) + 2 K_{o2} M_X(s))))$$

Charge transfer resistance

$$R_{ct} = \frac{1}{f F (O(0) K_{o1} \alpha_{o1} + X(0) (K_{o2} \alpha_{o2} + K_{r1} \alpha_{r1}) + R(0) K_{r2} \alpha_{r2})}$$

Concentration impedances

$$Z_O(s) = R_{ct} K_{o1} M_O(s) (O(0) K_{o1} \alpha_{o1} (1 + K_{r2} M_R(s)) + K_{o2} M_X(s)) + K_{r1} (X(0) \alpha_{r1} (1 + K_{r2} M_R(s)) + X(0) K_{o2} (\alpha_{o2} + \alpha_{r1}) M_X(s) + R(0) K_{r2} \alpha_{r2} M_X(s)) / (X(0) K_{r1} \alpha_{r1} (1 + K_{r2} M_R(s)) + 2 K_{o2} M_X(s)) + X(0) K_{o2} \alpha_{o2} (1 + 2 K_{r1} M_X(s)) + R(0) K_{r2} \alpha_{r2} (1 + 2 K_{r1} M_X(s)) + K_{o1} ((X(0) K_{o2} \alpha_{o2} + R(0) K_{r2} \alpha_{r2}) M_O(s) + O(0) \alpha_{o1} (1 + K_{r2} M_R(s) + 2 K_{o2} M_X(s)))$$

$$\begin{aligned}
Z_X(s) = & R_{ct} (K_{o2} - K_{r1}) M_X(s) (X(0) K_{o2} \alpha_{o2} - X(0) K_{r1} \alpha_{r1} (1 + K_{r2} M_R(s)) \\
& + R(0) K_{r2} \alpha_{r2} + K_{o1} ((X(0) K_{o2} \alpha_{o2} + R(0) K_{r2} \alpha_{r2}) M_O(s) - \alpha_{o1} O(0) (1 + K_{r2} M_R(s)))) / \\
& (X(0) K_{r1} \alpha_{r1} (1 + K_{r2} M_R(s) + 2 K_{o2} M_X(s)) \\
& + X(0) K_{o2} \alpha_{o2} (1 + 2 K_{r1} M_X(s)) + R(0) K_{r2} \alpha_{r2} (1 + 2 K_{r1} M_X(s)) \\
& + K_{o1} ((X(0) K_{o2} \alpha_{o2} + R(0) K_{r2} \alpha_{r2}) M_O(s) + O(0) \alpha_{o1} (1 + K_{r2} M_R(s) + 2 K_{o2} M_X(s))))
\end{aligned}$$

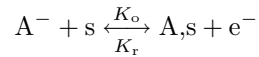
$$\begin{aligned}
Z_R(s) = & R_{ct} K_{r2} M_R(s) (R(0) K_{r2} \alpha_{r2} (1 + K_{o1} M_O(s) + K_{r1} M_X(s)) \\
& + K_{o2} (X(0) \alpha_{o2} + X(0) K_{r1} (\alpha_{o2} + \alpha_{r1}) M_X(s) + K_{o1} (X(0) \alpha_{o2} M_O(s) + O(0) \alpha_{o1} M_X(s)))) / \\
& (X(0) K_{r1} \alpha_{r1} (1 + K_{r2} M_R(s) + 2 K_{o2} M_X(s)) \\
& + X(0) K_{o2} \alpha_{o2} (1 + 2 K_{r1} M_X(s)) + R(0) K_{r2} \alpha_{r2} (1 + 2 K_{r1} M_X(s)) \\
& + K_{o1} ((X(0) K_{o2} \alpha_{o2} + R(0) K_{r2} \alpha_{r2}) M_O(s) + O(0) \alpha_{o1} (1 + K_{r2} M_R(s) + 2 K_{o2} M_X(s))))
\end{aligned}$$

## Chapter 2

# Reactions involving one adsorbate

### 2.1 Electroadsorption reaction (EAR)

#### 2.1.1 Mechanism



#### 2.1.2 Kinetic equations

No mass transport limitations, Langmuir isotherm

$$A^-(0, t) \approx A^{-*}, K_o = k_o A^{-*} \exp(\alpha_o f E), K_r = k_r \exp(-\alpha_r f E)$$

#### Transformation rates

$$v_{A^-}(t) = -v_1(t), v_s(t) = -v_1(t), v_A(t) = v_1(t)$$

#### Mass balance equations

Rate of production of adsorbed species

$$\frac{d\theta_s(t)}{dt} = \frac{v_s(t)}{\Gamma}, \frac{d\theta_A(t)}{dt} = \frac{v_A(t)}{\Gamma}$$

#### Current density vs. reaction rate

$$i_f(t) = F v(t)$$

#### Reaction rate

$$v(t) = \theta_s(t) \Gamma K_o(t) - \theta_A(t) \Gamma K_r(t)$$

### 2.1.3 Steady-state conditions

#### Steady-state equations

Adsorbed species

$$d\theta_s/dt = 0, \theta_A + \theta_s = 1$$

#### Steady-state solutions

Adsorbed species

$$\theta_s = \frac{K_r}{K_o + K_r}, \theta_A = \frac{K_o}{K_o + K_r}$$

Current density

$$i_f = 0$$

### 2.1.4 Faradaic impedance

Faradaic impedance

$$Z_f(s) = R_{ct} + Z_A(s) + Z_s(s)$$

$$Z_f(s) = \frac{s + K_o + K_r}{f F \Gamma s (\theta_s K_o \alpha_o + \theta_A K_r \alpha_r)} = \frac{(K_o + K_r) (s + K_o + K_r)}{f F s \Gamma K_o K_r}$$

Charge transfer resistance

$$R_{ct} = \frac{1}{f F \Gamma (\theta_s K_o \alpha_o + \theta_A K_r \alpha_r)} = \frac{K_o + K_r}{f F \Gamma K_o K_r}$$

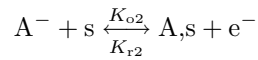
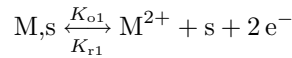
Concentration impedances

Adsorbed species

$$Z_A(s) = \frac{\Gamma K_r R_{ct}}{s} = \frac{K_o + K_r}{f F s \Gamma K_o}, Z_s(s) = \frac{K_o R_{ct}}{s} = \frac{K_o + K_r}{f F s \Gamma K_r}$$

## 2.2 Dissolution-passivation reaction

### 2.2.1 Mechanism [7]



### 2.2.2 Kinetic equations

No mass transport limitations, Langmuir isotherm

$$M^{2+}(0, t) \approx M^{2+*}, A^-(0, t) \approx A^{-*}$$

$$K_{o1} = k_{o1} \exp(2 \alpha_{o1} f E), K_{r1} = k_{r1} M^{2+*} \exp(-2 \alpha_{r1} f E)$$

$$K_{o2} = k_{o2} A^{-*} \exp(\alpha_{o2} f E), K_{r2} = k_{r2} \exp(-\alpha_{r2} f E)$$

**Transformation rates** ( $v_A$  stands for  $v_{A,s}$ )

$$v_s(t) = -v_2(t), v_A(t) = v_2(t)$$

**Mass balance equations**

Rate of production of adsorbed species

$$\frac{d\theta_s(t)}{dt} = \frac{v_s(t)}{\Gamma}, \frac{d\theta_A(t)}{dt} = \frac{v_A(t)}{\Gamma}$$

**Current density vs. step rates**

$$i_f(t) = F (2 v_1(t) + v_2(t))$$

**Step rates**

$$v_1(t) = \theta_s(t) \Gamma K_{o1}(t) - \theta_s(t) \Gamma K_{r1}(t), v_2(t) = \theta_s(t) \Gamma K_{o2}(t) - \theta_A(t) \Gamma K_{r2}(t)$$

### 2.2.3 Steady-state conditions

**Steady-state equations**

Adsorbed species

$$d\theta_s/dt = 0, \theta_A + \theta_s = 1$$

**Steady-state solutions**

Adsorbed species

$$\theta_s = \frac{K_{o2}}{K_{o2} + K_{r2}}, \theta_A = \frac{K_{r2}}{K_{o2} + K_{r2}}$$

Current density

$$i_f = \frac{2 F \Gamma (K_{o1} - K_{r1}) K_{r2}}{K_{o2} + K_{r2}}$$

### 2.2.4 Faradaic impedance

**Faradaic impedance**

Faradaic impedance

$$Z_f(s) = R_{ct} + Z_A(s) + Z_s(s)$$

$$Z_f(s) = (s + K_{o2} + K_{r2}) / (f F \Gamma (\theta_s (K_{o2} (s + 2 K_{r1}) \alpha_{o2} + 2 K_{o1} (2 (s + K_{o2} + K_{r2}) \alpha_{o1} - K_{o2} \alpha_{o2}) + 4 K_{r1} (s + K_{o2} + K_{r2}) \alpha_{r1}) + \theta_A (s - 2 K_{o1} + 2 K_{r1}) K_{r2} \alpha_{r2}))$$

$$Z_f(s) =$$

$$\frac{(K_{o2} + K_{r2}) (s + K_{o2} + K_{r2})}{f F \Gamma K_{r2} (4 (s + K_{r2}) (K_{o1} \alpha_{o1} + K_{r1} \alpha_{r1}) + K_{o2} (s + K_{o1} (-2 + 4 \alpha_{o1}) + K_{r1} (2 + 4 \alpha_{r1})))}$$

Charge transfer resistance

$$R_{ct} = \frac{1}{f F \Gamma (4 \theta_s K_{o1} \alpha_{o1} + \theta_s K_{o2} \alpha_{o2} + 4 \theta_s K_{r1} \alpha_{r1} + \theta_A K_{r2} \alpha_{r2})}$$

$$R_{ct} = \frac{K_{o2} + K_{r2}}{f F \Gamma K_{r2} (K_{o2} + 4 K_{o1} \alpha_{o1} + 4 K_{r1} \alpha_{r1})}$$

Concentration impedances

$$Z_A(s) = K_{r2} R_{ct} (\theta_s K_{o2} \alpha_{o2} + \theta_A K_{r2} \alpha_{r2}) /$$

$$(\theta_s (K_{o2} (s + 2 K_{r1}) \alpha_{o2} + 2 K_{o1} (2 (s + K_{o2} + K_{r2}) \alpha_{o1} - K_{o2} \alpha_{o2}) +$$

$$4 K_{r1} (s + K_{o2} + K_{r2}) \alpha_{r1}) + \theta_A (s - 2 K_{o1} + 2 K_{r1}) K_{r2} \alpha_{r2})$$

$$Z_A(s) = \frac{K_{o2} K_{r2} R_{ct}}{4 (s + K_{r2}) (K_{o1} \alpha_{o1} + K_{r1} \alpha_{r1}) + K_{o2} (s + K_{o1} (-2 + 4 \alpha_{o1}) + K_{r1} (2 + 4 \alpha_{r1}))}$$

$$Z_s(s) = -(2 K_{o1} + K_{o2} - 2 K_{r1}) R_{ct} (\theta_s K_{o2} \alpha_{o2} + \theta_A K_{r2} \alpha_{r2}) /$$

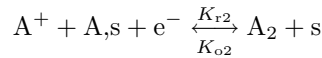
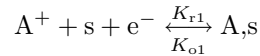
$$(\theta_s (-4 K_{o1} (s + K_{o2} + K_{r2}) \alpha_{o1} - K_{o2} (s - 2 K_{o1} + 2 K_{r1}) \alpha_{o2} -$$

$$4 K_{r1} (s + K_{o2} + K_{r2}) \alpha_{r1}) - \theta_A (s - 2 K_{o1} + 2 K_{r1}) K_{r2} \alpha_{r2})$$

$$Z_s(s) = \frac{K_{o2} (2 K_{o1} + K_{o2} - 2 K_{r1}) R_{ct}}{4 (s + K_{r2}) (K_{o1} \alpha_{o1} + K_{r1} \alpha_{r1}) + K_{o2} (s + K_{o1} (-2 + 4 \alpha_{o1}) + K_{r1} (2 + 4 \alpha_{r1}))}$$

## 2.3 Volmer-Heyrovský (V-H) reaction

### 2.3.1 Mechanism



### 2.3.2 Kinetic equations

No mass transport limitations, Langmuir isotherm

$$A^+(0, t) \approx A^{+*}, A_2(0, t) \approx A_2^*$$

$$K_{r1} = k_{r1} A^{+*} \exp(-\alpha_{r1} f E), K_{o1} = k_{o1} \exp(\alpha_{o1} f E)$$

$$K_{r2} = k_{r2} A^{+*} \exp(-\alpha_{r2} f E), K_{o2} = k_{o2} A_2^* \exp(\alpha_{o2} f E)$$

Transformation rates

$$v_s(t) = -v_1(t) + v_2(t), v_A(t) = v_1(t) - v_2(t)$$

## Mass balance equations

Rate of production of adsorbed species

$$\frac{d\theta_s(t)}{dt} = \frac{v_s(t)}{\Gamma}, \quad \frac{d\theta_A(t)}{dt} = \frac{v_A(t)}{\Gamma}$$

## Current density vs. step rates

$$i_f(t) = -F (v_1(t) + v_2(t))$$

## Step rates

$$v_1(t) = -\theta_A(t) \Gamma K_{o1}(t) + \theta_s(t) \Gamma K_{r1}(t), \quad v_2(t) = -\theta_s(t) \Gamma K_{o2}(t) + \theta_A(t) \Gamma K_{r2}(t)$$

### 2.3.3 Steady-state conditions

#### Steady-state equations

Adsorbed species

$$d\theta_s/dt = 0, \quad \theta_A + \theta_s = 1$$

#### Steady-state solutions

Adsorbed species

$$\theta_s = \frac{K_{o1} + K_{r2}}{K_{o1} + K_{o2} + K_{r1} + K_{r2}}, \quad \theta_A = \frac{K_{o2} + K_{r1}}{K_{o1} + K_{o2} + K_{r1} + K_{r2}}$$

Current density

$$i_f = \frac{2 F \Gamma (K_{o1} K_{o2} - K_{r1} K_{r2})}{K_{o1} + K_{o2} + K_{r1} + K_{r2}}$$

### 2.3.4 Faradaic impedance

#### Faradaic impedance

Faradaic impedance

$$Z_f(s) = R_{ct} + Z_A(s) + Z_s(s)$$

Charge transfer resistance

$$R_{ct} = \frac{1}{f F \Gamma (\theta_A K_{o1} \alpha_{o1} + \theta_s K_{o2} \alpha_{o2} + \theta_s K_{r1} \alpha_{r1} + \theta_A K_{r2} \alpha_{r2})}$$

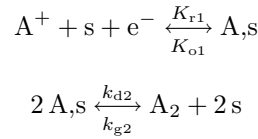
Concentration impedances

$$Z_A(s) = (K_{o1} - K_{r2}) R_{ct} (\theta_A K_{o1} \alpha_{o1} - \theta_s K_{o2} \alpha_{o2} + \theta_s K_{r1} \alpha_{r1} - \theta_A K_{r2} \alpha_{r2}) / \\ (\theta_s K_{o2} (s + 2 K_{r1}) \alpha_{o2} + \theta_s K_{r1} (s + 2 K_{o2} + 2 K_{r2}) \alpha_{r1} + \theta_A (s + 2 K_{r1}) K_{r2} \alpha_{r2} + \\ K_{o1} (\theta_A (s + 2 K_{o2} + 2 K_{r2}) \alpha_{o1} + 2 (\theta_s K_{o2} \alpha_{o2} + \theta_A K_{r2} \alpha_{r2})))$$

$$Z_s(s) = (K_{o2} - K_{r1}) R_{ct} (-\theta_A K_{o1} \alpha_{o1} + \theta_s K_{o2} \alpha_{o2} - \theta_s K_{r1} \alpha_{r1} + \theta_A K_{r2} \alpha_{r2}) / (\theta_s K_{o2} (s + 2 K_{r1}) \alpha_{o2} + \theta_s K_{r1} (s + 2 K_{o2} + 2 K_{r2}) \alpha_{r1} + \theta_A (s + 2 K_{r1}) K_{r2} \alpha_{r2} + K_{o1} (\theta_A (s + 2 K_{o2} + 2 K_{r2}) \alpha_{o1} + 2 (\theta_s K_{o2} \alpha_{o2} + \theta_A K_{r2} \alpha_{r2})))$$

## 2.4 Volmer-Tafel (V-T) reaction

### 2.4.1 Mechanism



### 2.4.2 Kinetic equations

No mass transport limitations, Langmuir isotherm

$$A^+(0, t) \approx A^{+*}, \quad A_2(0, t) \approx A_2^*$$

$$K_{r1} = k_{r1} A^{+*} \exp(-\alpha_{r1} f E), \quad K_{o1} = k_{o1} \exp(\alpha_{o1} f E), \quad k_{g2} = k'_{g2} A_2^*$$

#### Transformation rates

$$v_s(t) = -v_1(t) + 2 v_2(t), \quad v_A(t) = v_1(t) - 2 v_2(t)$$

#### Mass balance equations

Rate of production of adsorbed species

$$\frac{d\theta_s(t)}{dt} = \frac{v_s(t)}{\Gamma}, \quad \frac{d\theta_A(t)}{dt} = \frac{v_A(t)}{\Gamma}$$

#### Current density vs. step rates

$$i_f(t) = -F v_1(t)$$

#### Step rates

$$v_1(t) = -\theta_A(t) \Gamma K_{o1}(t) + \theta_s(t) \Gamma K_{r1}(t), \quad v_2(t) = \theta_A(t)^2 \Gamma^2 k_{d2} - \theta_s(t)^2 \Gamma^2 k_{g2}$$

### 2.4.3 Steady-state conditions

#### Steady-state equations

Adsorbed species

$$d\theta_s/dt = 0, \quad \theta_A + \theta_s = 1$$



## Steady-state solutions

Adsorbed species

$$\theta_A = \frac{-\left(4\Gamma k_{g2} + K_{o1} + K_{r1} - \sqrt{8\Gamma k_{g2} K_{o1} + 8\Gamma k_{d2} (2\Gamma k_{g2} + K_{r1}) + (K_{o1} + K_{r1})^2}\right)}{4\Gamma (k_{d2} - k_{g2})}$$

Current density

$$i_f = \frac{F}{4(k_{d2} - k_{g2})} \left( -\left(4\Gamma k_{g2} K_{o1} + 4\Gamma k_{d2} K_{r1} + (K_{o1} + K_{r1})^2\right) + \right. \\ \left. (K_{o1} + K_{r1}) \sqrt{8\Gamma k_{g2} K_{o1} + 8\Gamma k_{d2} (2\Gamma k_{g2} + K_{r1}) + (K_{o1} + K_{r1})^2} \right)$$

## 2.4.4 Faradaic impedance

Faradaic impedance

Faradaic impedance

$$Z_f(s) = R_{ct} + Z_A(s) + Z_s(s)$$

$$Z_f(s) = \frac{s + 4\theta_A \Gamma k_{d2} + 4\theta_s \Gamma k_{g2} + K_{o1} + K_{r1}}{f F \Gamma (s + 4\theta_A \Gamma k_{d2} + 4\theta_s \Gamma k_{g2}) (\theta_A K_{o1} \alpha_{o1} + \theta_s K_{r1} \alpha_{r1})}$$

Charge transfer resistance

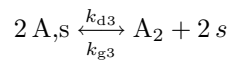
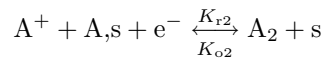
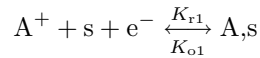
$$R_{ct} = \frac{1}{f F \Gamma (\theta_A K_{o1} \alpha_{o1} + \theta_s K_{r1} \alpha_{r1})}$$

Concentration impedances

$$Z_A(s) = \frac{K_{o1} R_{ct}}{s + 4\theta_A \Gamma k_{d2} + 4\theta_s \Gamma k_{g2}}, \quad Z_s(s) = \frac{K_{r1} R_{ct}}{s + 4\theta_A \Gamma k_{d2} + 4\theta_s \Gamma k_{g2}}$$

## 2.5 Volmer-Heyrovský-Tafel (V-H-T) reaction

### 2.5.1 Mechanism



## 2.5.2 Kinetic equations

No mass transport limitations, Langmuir isotherm

$$A^+(0, t) \approx A^{+*}, A_2(0, t) \approx A_2^*$$

$$K_{r1} = k_{r1} A^{+*} \exp(-\alpha_{r1} f E), K_{o1} = k_{o1} \exp(\alpha_{o1} f E)$$

$$K_{r2} = k_{r2} A^{+*} \exp(-\alpha_{r2} f E), K_{o2} = k_{o2} A_2^* \exp(\alpha_{o2} f E), k_{g3} = k'_{g3} A_2^*$$

### Transformation rates

$$v_s(t) = -v_1(t) + v_2(t) + 2v_3(t), v_A(t) = v_1(t) - v_2(t) - 2v_3(t)$$

### Mass balance equations

Rate of production of adsorbed species

$$\frac{d\theta_s(t)}{dt} = \frac{v_s(t)}{\Gamma}, \frac{d\theta_A(t)}{dt} = \frac{v_A(t)}{\Gamma}$$

### Current density vs. step rates

$$i_f(t) = -F (v_1(t) + v_2(t))$$

### Step rates

$$v_1(t) = -\theta_A(t) \Gamma K_{o1}(t) + \theta_s(t) \Gamma K_{r1}(t)$$

$$v_2(t) = -\theta_s(t) \Gamma K_{o2}(t) + \theta_A(t) \Gamma K_{r2}(t)$$

$$v_3(t) = \theta_A(t)^2 \Gamma^2 k_{d3} - \theta_s(t)^2 \Gamma^2 k_{g3}$$

## 2.5.3 Steady-state conditions

### Steady-state equations

Adsorbed species

$$d\theta_s/dt = 0, \theta_A + \theta_s = 1$$

### Steady-state solutions

Adsorbed species

$$\theta_A = \frac{1}{4\Gamma (k_{g3} - k_{d3})} \left( 4\Gamma k_{g3} + K_{o1} + K_{o2} + K_{r1} + K_{r2} - \sqrt{8\Gamma (k_{d3} - k_{g3}) (2\Gamma k_{g3} + K_{o2} + K_{r1}) + (4\Gamma k_{g3} + K_{o1} + K_{o2} + K_{r1} + K_{r2})^2} \right)$$

Current density

$$i_f = \frac{F}{4(k_{d3} - k_{g3})} \left( 4\Gamma k_{d3} (K_{o2} - K_{r1}) - (K_{o1} + K_{r1})^2 + 4\Gamma k_{g3} (-K_{o1} + K_{r2}) + (K_{o2} + K_{r2})^2 + (K_{o1} - K_{o2} + K_{r1} - K_{r2}) \right) \times \sqrt{8\Gamma (k_{d3} - k_{g3}) (2\Gamma k_{g3} + K_{o2} + K_{r1}) + (4\Gamma k_{g3} + K_{o1} + K_{o2} + K_{r1} + K_{r2})^2}$$

## 2.5.4 Faradaic impedance

Faradaic impedance

$$Z_f(s) = R_{ct} + Z_A(s) + Z_s(s)$$

$$\begin{aligned} Z_f(s) = & (s + 4\theta_A \Gamma k_{d3} + 4\theta_s \Gamma k_{g3} + K_{o1} + K_{o2} + K_{r1} + K_{r2}) / \\ & (f F \Gamma (\theta_s K_{o2} (s + 4\theta_A \Gamma k_{d3} + 4\theta_s \Gamma k_{g3} + 2K_{r1}) \alpha_{o2} \\ & + \theta_s K_{r1} (s + 4\theta_A \Gamma k_{d3} + 4\theta_s \Gamma k_{g3} + 2K_{o2} + 2K_{r2}) \alpha_{r1} \\ & + \theta_A (s + 4\theta_A \Gamma k_{d3} + 4\theta_s \Gamma k_{g3} + 2K_{r1}) K_{r2} \alpha_{r2} \\ & + K_{o1} (\theta_A (s + 4\theta_A \Gamma k_{d3} + 4\theta_s \Gamma k_{g3} + 2K_{o2} + 2K_{r2}) \alpha_{o1} + 2 (\theta_s K_{o2} \alpha_{o2} + \theta_A K_{r2} \alpha_{r2}))) \end{aligned}$$

Charge transfer resistance

$$R_{ct} = \frac{1}{f F \Gamma (\theta_A K_{o1} \alpha_{o1} + \theta_s K_{o2} \alpha_{o2} + \theta_s K_{r1} \alpha_{r1} + \theta_A K_{r2} \alpha_{r2})}$$

Concentration impedances

$$\begin{aligned} Z_A(s) = & (K_{o1} - K_{r2}) R_{ct} (\theta_A K_{o1} \alpha_{o1} - \theta_s K_{o2} \alpha_{o2} + \theta_s K_{r1} \alpha_{r1} - \theta_A K_{r2} \alpha_{r2}) / \\ & (\theta_s K_{o2} (s + 4\theta_A \Gamma k_{d3} + 4\theta_s \Gamma k_{g3} + 2K_{r1}) \alpha_{o2} \\ & + \theta_s K_{r1} (s + 4\theta_A \Gamma k_{d3} + 4\theta_s \Gamma k_{g3} + 2K_{o2} + 2K_{r2}) \alpha_{r1} \\ & + \theta_A (s + 4\theta_A \Gamma k_{d3} + 4\theta_s \Gamma k_{g3} + 2K_{r1}) K_{r2} \alpha_{r2} \\ & + K_{o1} (\theta_A (s + 4\theta_A \Gamma k_{d3} + 4\theta_s \Gamma k_{g3} + 2K_{o2} + 2K_{r2}) \alpha_{o1} + 2 (\theta_s K_{o2} \alpha_{o2} + \theta_A K_{r2} \alpha_{r2}))) \end{aligned}$$

$$\begin{aligned} Z_s(s) = & (K_{o2} - K_{r1}) R_{ct} (-\theta_A K_{o1} \alpha_{o1} + \theta_s K_{o2} \alpha_{o2} - \theta_s K_{r1} \alpha_{r1} + \theta_A K_{r2} \alpha_{r2}) / \\ & (\theta_s K_{o2} (s + 4\theta_A \Gamma k_{d3} + 4\theta_s \Gamma k_{g3} + 2K_{r1}) \alpha_{o2} \\ & + \theta_s K_{r1} (s + 4\theta_A \Gamma k_{d3} + 4\theta_s \Gamma k_{g3} + 2K_{o2} + 2K_{r2}) \alpha_{r1} \\ & + \theta_A (s + 4\theta_A \Gamma k_{d3} + 4\theta_s \Gamma k_{g3} + 2K_{r1}) K_{r2} \alpha_{r2} \\ & + K_{o1} (\theta_A (s + 4\theta_A \Gamma k_{d3} + 4\theta_s \Gamma k_{g3} + 2K_{o2} + 2K_{r2}) \alpha_{o1} + 2 (\theta_s K_{o2} \alpha_{o2} + \theta_A K_{r2} \alpha_{r2}))) \end{aligned}$$

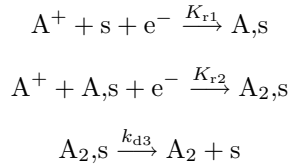


# Chapter 3

## Reactions involving two adsorbates

### 3.1 Volmer-Heyrovský with chemical desorption

#### 3.1.1 Mechanism [6, 3, 4]



#### 3.1.2 Kinetic equations

No mass transfer limitations, Langmuir isotherm

$$A^+(0, t) \approx A^{+*}$$

$$K_{r1} = k_{r1} A^{+*} \exp(-\alpha_{r1} f E), \quad K_{r2} = k_{r2} A^{+*} \exp(-\alpha_{r2} f E)$$

#### Transformation rates

$$v_s(t) = -v_1(t) + v_3(t), \quad v_A(t) = v_1(t) - v_2(t), \quad v_{A_2}(t) = v_2(t) - v_3(t)$$

#### Mass balance equations

Rate of production of adsorbed species

$$\frac{d\theta_s(t)}{dt} = \frac{v_s(t)}{\Gamma}, \quad \frac{d\theta_A(t)}{dt} = \frac{v_A(t)}{\Gamma}, \quad \frac{d\theta_{A_2}(t)}{dt} = \frac{v_{A_2}(t)}{\Gamma}$$

#### Current density vs. step rates

$$i_f(t) = -F (v_1(t) + v_2(t))$$

### Step rates

$$v_1(t) = \theta_s(t) \Gamma K_{r1}(t), v_2(t) = \theta_A(t) \Gamma K_{r2}(t), v_3(t) = \theta_{A_2}(t) \Gamma k_{d3}$$

### 3.1.3 Steady-state conditions

#### Steady-state equations

Adsorbed species

$$d\theta_s/dt = 0, d\theta_A/dt = 0, \theta_A + \theta_{A_2} + \theta_s = 1$$

#### Steady-state solutions

Adsorbed species

$$\theta_A = \frac{k_{d3} K_{r1}}{K_{r1} K_{r2} + k_{d3} (K_{r1} + K_{r2})}, \theta_{A_2} = \frac{K_{r1} K_{r2}}{K_{r1} K_{r2} + k_{d3} (K_{r1} + K_{r2})}$$

Current density

$$i_f = \frac{-2 F \Gamma k_{d3} K_{r1} K_{r2}}{K_{r1} K_{r2} + k_{d3} (K_{r1} + K_{r2})}$$

### 3.1.4 Faradaic impedance

Faradaic impedance

$$Z_f(s) = R_{ct} + Z_A(s) + Z_s(s)$$

Charge transfer resistance

$$R_{ct} = \frac{1}{f F \Gamma (\theta_s K_{r1} \alpha_{r1} + \theta_A K_{r2} \alpha_{r2})}$$

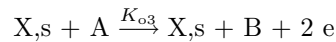
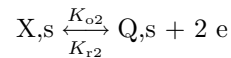
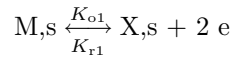
Concentration impedances

$$Z_A(s) = \frac{K_{r2} R_{ct} (-\theta_s (s + k_{d3}) K_{r1} \alpha_{r1}) + \theta_A (s + k_{d3} + K_{r1}) K_{r2} \alpha_{r2}}{\theta_s (s + k_{d3}) K_{r1} (s + 2 K_{r2}) \alpha_{r1} + \theta_A (s (s + k_{d3}) + (s + 2 k_{d3}) K_{r1}) K_{r2} \alpha_{r2}}$$

$$Z_s(s) = \frac{K_{r1} R_{ct} (\theta_s K_{r1} (s + K_{r2}) \alpha_{r1} - k_{d3} (-\theta_s K_{r1} \alpha_{r1} + \theta_A K_{r2} \alpha_{r2}))}{\theta_s (s + k_{d3}) K_{r1} (s + 2 K_{r2}) \alpha_{r1} + \theta_A (s (s + k_{d3}) + (s + 2 k_{d3}) K_{r1}) K_{r2} \alpha_{r2}}$$

## 3.2 Schuhmann dissolution-passivation reaction # 1

### 3.2.1 Mechanism [7]



### 3.2.2 Kinetic equations

No mass transfer limitations, Langmuir isotherm

$$A(0, t) \approx A^*$$

$$K_{o1} = k_{o1} \exp(2 \alpha_{o1} f E), K_{r1} = k_{r1} \exp(-2 \alpha_{r1} f E)$$

$$K_{o2} = k_{o2} \exp(2 \alpha_{o2} f E), K_{r2} = k_{r2} \exp(-2 \alpha_{r2} f E), K_{o3} = k_{o3} \exp(2 \alpha_{o3} f E)$$

#### Transformation rates

$$v_s(t) = -v_1(t), v_X(t) = v_1(t) - v_2(t), v_Q(t) = v_2(t)$$

#### Mass balance equations

Rate of production of adsorbed species

$$\frac{d\theta_s(t)}{dt} = \frac{v_s(t)}{\Gamma}, \frac{d\theta_X(t)}{dt} = \frac{v_X(t)}{\Gamma}, \frac{d\theta_Q(t)}{dt} = \frac{v_Q(t)}{\Gamma}$$

#### Current density vs. step rates

$$i_f(t) = 2 F (v_1(t) + v_2(t) + v_3(t))$$

#### Step rates

$$v_1(t) = \theta_s(t) \Gamma K_{o1}(t) - \theta_X(t) \Gamma K_{r1}(t)$$

$$v_2(t) = \theta_X(t) \Gamma K_{o2}(t) - \theta_Q(t) \Gamma K_{r2}(t)$$

$$v_3(t) = \theta_X(t) \Gamma K_{o3}(t)$$

### 3.2.3 Steady-state conditions

#### Steady-state equations

Adsorbed species

$$d\theta_s/dt = 0, d\theta_X/dt = 0, \theta_Q + \theta_s + \theta_X = 1$$

#### Steady-state solutions

Adsorbed species

$$\theta_Q = \frac{K_{o1} K_{o2}}{K_{r1} K_{r2} + K_{o1} (K_{o2} + K_{r2})}, \theta_X = \frac{K_{o1} K_{r2}}{K_{r1} K_{r2} + K_{o1} (K_{o2} + K_{r2})}$$

Current density

$$i_f = \frac{2 F \Gamma K_{o1} K_{o3} K_{r2}}{K_{r1} K_{r2} + K_{o1} (K_{o2} + K_{r2})}$$

### 3.2.4 Faradaic impedance

Faradaic impedance

$$Z_f(s) = R_{ct} + Z_Q(s) + Z_s(s) + Z_X(s)$$

Charge transfer resistance

$$R_{ct} = \frac{1}{4 f F \Gamma (\theta_s K_{o1} \alpha_{o1} + \theta_X K_{o2} \alpha_{o2} + \theta_X K_{o3} \alpha_{o3} + \theta_X K_{r1} \alpha_{r1} + \theta_Q K_{r2} \alpha_{r2})}$$

Concentration impedances

$$\begin{aligned} Z_Q(s) = & K_{r2} R_{ct} (K_{o2} (\theta_X (s + K_{r1}) \alpha_{o2} + K_{o1} (\theta_s \alpha_{o1} + \theta_X \alpha_{o2}) + \\ & \theta_X K_{r1} \alpha_{r1}) + \theta_Q (s + K_{o1} + K_{r1}) K_{r2} \alpha_{r2}) / \\ & (\theta_X s^2 K_{o3} \alpha_{o3} + \theta_X s K_{o3} K_{r1} \alpha_{o3} + \theta_X s K_{o3} K_{r2} \alpha_{o3} + \theta_X K_{o3} K_{r1} K_{r2} \alpha_{o3} + \\ & \theta_X s^2 K_{r1} \alpha_{r1} + \theta_X s K_{o3} K_{r1} \alpha_{r1} + \theta_X s K_{r1} K_{r2} \alpha_{r1} + \theta_X K_{o3} K_{r1} K_{r2} \alpha_{r1} + \\ & \theta_X s K_{o2} ((s - K_{o3} + 2 K_{r1}) \alpha_{o2} + K_{o3} \alpha_{o3} + 2 K_{r1} \alpha_{r1}) + \\ & \theta_Q s^2 K_{r2} \alpha_{r2} - \theta_Q s K_{o3} K_{r2} \alpha_{r2} + 2 \theta_Q s K_{r1} K_{r2} \alpha_{r2} + \\ & K_{o1} (\theta_s (2 s K_{o2} + (s + K_{o3}) (s + K_{r2})) \alpha_{o1} + \theta_X s K_{o3} \alpha_{o3} + \\ & \theta_X K_{o3} K_{r2} \alpha_{o3} + \theta_X K_{o2} ((s - K_{o3}) \alpha_{o2} + K_{o3} \alpha_{o3}) + \theta_Q s K_{r2} \alpha_{r2} - \theta_Q K_{o3} K_{r2} \alpha_{r2})) \end{aligned}$$

$$\begin{aligned} Z_s(s) = & K_{o1} R_{ct} (\theta_s K_{o1} (s + K_{o2} + K_{r2}) \alpha_{o1} + \\ & K_{r1} (\theta_X K_{o2} \alpha_{o2} + \theta_X (s + K_{o2} + K_{r2}) \alpha_{r1} + \theta_Q K_{r2} \alpha_{r2})) / \\ & (\theta_X s^2 K_{o3} \alpha_{o3} + \theta_X s K_{o3} K_{r1} \alpha_{o3} + \theta_X s K_{o3} K_{r2} \alpha_{o3} + \theta_X K_{o3} K_{r1} K_{r2} \alpha_{o3} + \\ & \theta_X s^2 K_{r1} \alpha_{r1} + \theta_X s K_{o3} K_{r1} \alpha_{r1} + \theta_X s K_{r1} K_{r2} \alpha_{r1} + \theta_X K_{o3} K_{r1} K_{r2} \alpha_{r1} + \\ & \theta_X s K_{o2} ((s - K_{o3} + 2 K_{r1}) \alpha_{o2} + K_{o3} \alpha_{o3} + 2 K_{r1} \alpha_{r1}) + \\ & \theta_Q s^2 K_{r2} \alpha_{r2} - \theta_Q s K_{o3} K_{r2} \alpha_{r2} + 2 \theta_Q s K_{r1} K_{r2} \alpha_{r2} + \\ & K_{o1} (\theta_s (2 s K_{o2} + (s + K_{o3}) (s + K_{r2})) \alpha_{o1} + \theta_X s K_{o3} \alpha_{o3} + \\ & \theta_X K_{o3} K_{r2} \alpha_{o3} + \theta_X K_{o2} ((s - K_{o3}) \alpha_{o2} + K_{o3} \alpha_{o3}) + \theta_Q s K_{r2} \alpha_{r2} - \theta_Q K_{o3} K_{r2} \alpha_{r2})) \end{aligned}$$

$$\begin{aligned} Z_X(s) = & (K_{o2} + K_{o3} - K_{r1}) R_{ct} (\theta_X s K_{o2} \alpha_{o2} - \theta_X K_{r1} (s + K_{r2}) \alpha_{r1} + \theta_Q s K_{r2} \alpha_{r2} + \\ & K_{o1} (- (\theta_s (s + K_{r2}) \alpha_{o1}) + \theta_X K_{o2} \alpha_{o2} + \theta_Q K_{r2} \alpha_{r2})) / \\ & (\theta_X (s K_{o2} ((s - K_{o3} + 2 K_{r1}) \alpha_{o2} + K_{o3} \alpha_{o3} + 2 K_{r1} \alpha_{r1}) + \\ & (s + K_{r2}) (K_{o3} (s + K_{r1}) \alpha_{o3} + (s + K_{o3}) K_{r1} \alpha_{r1})) + \theta_Q s (s - K_{o3} + 2 K_{r1}) K_{r2} \alpha_{r2} + \\ & K_{o1} (\theta_s (2 s K_{o2} + (s + K_{o3}) (s + K_{r2})) \alpha_{o1} + \theta_X K_{o2} (s - K_{o3}) \alpha_{o2} + \\ & \theta_X K_{o3} (s + K_{o2} + K_{r2}) \alpha_{o3} + \theta_Q (s - K_{o3}) K_{r2} \alpha_{r2})) \end{aligned}$$

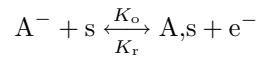


# Chapter 4

## Reactions involving both adsorbed and soluble species

### 4.1 Electroadsorption reaction (EAR) with limitation by mass transport

#### 4.1.1 Mechanism



#### 4.1.2 Kinetic equations

Langmuir isotherm :  $K_o = k_o \exp(\alpha_o f E)$ ,  $K_r = k_r \exp(-\alpha_r f E)$

#### Transformation rates

$$v_{A^-}(t) = -v_1(t), v_s(t) = -v_1(t), v_A(t) = v_1(t)$$

#### Mass balance equations

Flux of soluble species

$$J_{A^-}(0, t) = v_{A^-}(t)$$

Rate of production of adsorbed species

$$\frac{d\theta_s(t)}{dt} = \frac{v_s(t)}{\Gamma}, \frac{d\theta_A(t)}{dt} = \frac{v_A(t)}{\Gamma}$$

#### Current density vs. reaction rate

$$i_f(t) = F v(t)$$

### Reaction rate

$$v(t) = A^-(0, t) \theta_s(t) \Gamma K_o(t) - \theta_A(t) \Gamma K_r(t)$$

### 4.1.3 Steady-state conditions

#### Steady-state equations

Soluble species

$$J_{A^-}(0) = - \left( A^{-*} - A^-(0) \right) m_{A^-}$$

Adsorbed species

$$d\theta_s/dt = 0, \theta_A + \theta_s = 1$$

#### Steady-state solutions

Soluble species

$$A^-(0) = A^{-*}$$

Adsorbed species

$$\theta_s = \frac{K_r}{A^{-*} K_o + K_r}, \theta_A = \frac{A^{-*} K_o}{A^{-*} K_o + K_r}$$

Current density

$$i_f = 0$$

### 4.1.4 Faradaic impedance

Faradaic impedance

$$Z_f(s) = R_{ct} + Z_{A^-}(s) + Z_A(s) + Z_s(s)$$

Charge transfer resistance

$$R_{ct} = \frac{1}{f F \Gamma (A^{-*} \theta_s K_o \alpha_o + \theta_A K_r \alpha_r)}$$

Concentration impedances

Soluble species

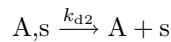
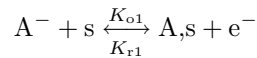
$$Z_{A^-}(s) = \theta_s \Gamma K_o R_{ct} M_{A^-}(s), M_{A^-}(s) = \frac{1}{m_{A^-}} \frac{\text{th} \sqrt{\tau_{A^-} s}}{\sqrt{\tau_{A^-} s}}$$

Adsorbed species

$$Z_s(s) = \frac{A^{-*} K_o R_{ct}}{s}, Z_A(s) = \frac{\Gamma K_r R_{ct}}{s}$$

## 4.2 Electrosorption-desorption reaction

### 4.2.1 Mechanism



## 4.2.2 Kinetic equations

Langmuir isotherm:  $K_{o1} = k_{o1} \exp(\alpha_{o1} f E)$ ,  $K_{r1} = k_{r1} \exp(-\alpha_{r1} f E)$

### Transformation rates

$$v_{A^-}(t) = -v_1(t), v_s(t) = -v_1(t) + v_2(t), v_A(t) = v_1(t) - v_2(t)$$

### Mass balance equations

Flux of soluble species

$$J_{A^-}(0, t) = v_{A^-}(t)$$

Rate of production of adsorbed species

$$\frac{d\theta_s(t)}{dt} = \frac{v_s(t)}{\Gamma}, \frac{d\theta_A(t)}{dt} = \frac{v_A(t)}{\Gamma}$$

### Current density vs. step rates

$$i_f(t) = F v_1(t)$$

### Step rates

$$v_1(t) = A^-(0, t) \theta_s(t) \Gamma K_{o1}(t) - \theta_A(t) \Gamma K_{r1}(t), v_2(t) = \theta_A(t) \Gamma k_{d2}$$

## 4.2.3 Steady-state conditions

### Steady-state equations

Soluble species

$$J_{A^-}(0) = - (A^{-*} - A^-(0)) m_{A^-}$$

Adsorbed species

$$d\theta_s/dt = 0, \theta_A + \theta_s = 1$$

### Steady-state solutions

Soluble species

$$A^-(0) = \frac{1}{2 K_{o1} m_{A^-}} \left( A^{-*} K_{o1} m_{A^-} - K_{r1} m_{A^-} - k_{d2} (\Gamma K_{o1} + m_{A^-}) + \sqrt{4 \Gamma k_{d2} K_{o1} (k_{d2} + K_{r1}) m_{A^-} + ((A^{-*} K_{o1} + K_{r1}) m_{A^-} + k_{d2} (-\Gamma K_{o1} + m_{A^-}))^2} \right)$$

Adsorbed species

$$\theta_A = \frac{1}{2 \Gamma k_{d2} K_{o1}} \left( A^{-*} K_{o1} m_{A^-} + K_{r1} m_{A^-} + k_{d2} (\Gamma K_{o1} + m_{A^-}) - \sqrt{4 \Gamma k_{d2} K_{o1} (k_{d2} + K_{r1}) m_{A^-} + ((sA^{-*} K_{o1} + K_{r1}) m_{A^-} + k_{d2} (-\Gamma K_{o1} + m_{A^-}))^2} \right)$$

Current density

$$i_f = \frac{F}{2K_{o1}} \left( \left( A^{-*} K_{o1} + K_{r1} \right) m_{A^-} + k_{d2} (\Gamma K_{o1} + m_{A^-}) - \sqrt{4\Gamma k_{d2} K_{o1} (k_{d2} + K_{r1}) m_{A^-} + \left( \left( A^{-*} K_{o1} + K_{r1} \right) m_{A^-} + k_{d2} (-\Gamma K_{o1} + m_{A^-}) \right)^2} \right)$$

#### 4.2.4 Faradaic impedance

Faradaic impedance

$$Z_f(s) = R_{ct} + Z_{A^-}(s) + Z_A(s) + Z_s(s)$$

$$Z_f(s) = \frac{s + k_{d2} + A^-(0) K_{o1} + K_{r1} + \theta_s \Gamma (s + k_{d2}) K_{o1} M_{A^-}(s)}{f F \Gamma (s + k_{d2}) (A^-(0) \theta_s K_{o1} \alpha_{o1} + \theta_A K_{r1} \alpha_{r1})}$$

$$M_{A^-}(s) = \frac{1}{m_{A^-}} \frac{\text{th} \sqrt{\tau_{A^-} s}}{\sqrt{\tau_{A^-} s}}$$

Charge transfer resistance

$$R_{ct} = \frac{1}{f F \Gamma (A^-(0) \theta_s K_{o1} \alpha_{o1} + \theta_A K_{r1} \alpha_{r1})}$$

Concentration impedances

Soluble species

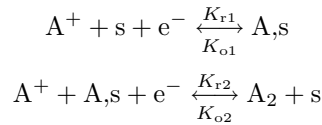
$$Z_{A^-}(s) = \theta_s \Gamma K_{o1} R_{ct} M_{A^-}(s)$$

Adsorbed species

$$Z_s(s) = \frac{A^-(0) K_{o1} R_{ct}}{s + k_{d2}}, Z_A(s) = \frac{K_{r1} R_{ct}}{s + k_{d2}}$$

### 4.3 (V-H) reaction with mass transport limitation

#### 4.3.1 Mechanism



#### 4.3.2 Kinetic equations

Langmuir isotherm

$$K_{r1} = k_{r1} \exp(-\alpha_{r1} f E), K_{o1} = k_{o1} \exp(\alpha_{o1} f E)$$

$$K_{r2} = k_{r2} \exp(-\alpha_{r2} f E), K_{o2} = k_{o2} \exp(\alpha_{o2} f E)$$

### Transformation rates

$$v_{A^+}(t) = -v_1(t) - v_2(t), v_s(t) = -v_1(t) + v_2(t), v_A(t) = v_1(t) - v_2(t)$$

### Mass balance equations

Flux of soluble species

$$J_{A^+}(0, t) = v_{A^+}(t), J_{A_2}(0, t) = v_{A_2}(t)$$

Rate of productions of adsorbed species

$$\frac{d\theta_s(t)}{dt} = \frac{v_s(t)}{\Gamma}, \frac{d\theta_A(t)}{dt} = \frac{v_A(t)}{\Gamma}$$

### Current density vs. step rate

$$i_f(t) = -F (v_1(t) + v_2(t))$$

### Step rates

$$v_1(t) = -\theta_A(t) \Gamma K_{o1}(t) + A^+(0, t) \theta_s(t) \Gamma K_{r1}(t)$$

$$v_2(t) = -A_2(0, t) \theta_s(t) \Gamma K_{o2}(t) + A^+(0, t) \theta_A(t) \Gamma K_{r2}(t)$$

## 4.3.3 Steady-state conditions

### Steady-state equations

Soluble species

$$J_{A^+}(0) = - (A^{+*} - A^+(0)) m_{A^+}, J_{A_2}(0) = - (A_2^* - A_2(0)) m_{A_2}$$

Adsorbed species

$$d\theta_s/dt = 0, \theta_A + \theta_s = 1$$

### Steady-state solution

Soluble species (<sup>1</sup>)

$$A^+(0) \rightarrow - (m_{A^+} (- (K_{r1} + K_{r2}) m_{A_2} A^{+*} + K_{o1} (\Gamma K_{o2} + m_{A_2}) + K_{o2} (m_{A^+} A^{+*} + A_2^* m_{A_2})) +$$

$$\sqrt{((4 \Gamma K_{o1} K_{r2} m_{A_2} + m_{A^+} (((K_{r1} + K_{r2}) A^{+*} + K_{o1} + A_2^* K_{o2}) m_{A_2} - \Gamma K_{o1} K_{o2}))^2 +$$

$$4 \Gamma (K_{r2} m_{A^+} A^{+*} + K_{o1} (2 \Gamma K_{r2} + m_{A^+})) m_{A_2} (K_{o2} (2 K_{r1} m_{A_2} A_2^* + (K_{r1} A^{+*} + K_{o1}) m_{A^+}) - 2 K_{o1} K_{r2} m_{A_2})) /$$

$$(-K_{o2} m_{A^+}^2 + 2 (K_{r1} + K_{r2}) m_{A_2} m_{A^+} + 4 \Gamma K_{r1} K_{r2} m_{A_2}))$$

$$A_2(0) \rightarrow (K_{r1} m_{A_2} (4 \Gamma K_{r2} (m_{A^+} A^{+*} + 2 A_2^* m_{A_2}) + m_{A^+} (m_{A^+} A^{+*} + 4 A_2^* m_{A_2})) +$$

$$m_{A^+} (\Gamma K_{o1} K_{o2} m_{A^+} + m_{A_2} (4 K_{r2} m_{A_2} A_2^* + (K_{r2} A^{+*} + K_{o1} - A_2^* K_{o2}) m_{A^+})) +$$

$$\sqrt{((4 \Gamma K_{o1} K_{r2} m_{A_2} + m_{A^+} (((K_{r1} + K_{r2}) A^{+*} + K_{o1} + A_2^* K_{o2}) m_{A_2} - \Gamma K_{o1} K_{o2}))^2 + 4 \Gamma (K_{r2} m_{A^+} A^{+*} +$$

$$K_{o1} (2 \Gamma K_{r2} + m_{A^+})) m_{A_2} (K_{o2} (2 K_{r1} m_{A_2} A_2^* + (K_{r1} A^{+*} + K_{o1}) m_{A^+}) - 2 K_{o1} K_{r2} m_{A_2})) /$$

$$(2 m_{A_2} (-K_{o2} m_{A^+}^2 + 2 (K_{r1} + K_{r2}) m_{A_2} m_{A^+} + 4 \Gamma K_{r1} K_{r2} m_{A_2}))$$

Adsorbed species

Current density

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<sup>1</sup> → stands for =.

$$\theta_A \rightarrow -\left(4\Gamma K_{o2} K_{r1} m_{A_2} A_2^* + m_{A^+} (\Gamma K_{o2} (2 K_{r1} A^{++} + K_{o1}) + ((K_{r1} + K_{r2}) A^{++} + K_{o1} + A_2^* K_{o2}) m_{A_2}) + \sqrt{\left((4\Gamma K_{o1} K_{r2} m_{A_2} + m_{A^+} (((K_{r1} + K_{r2}) A^{++} + K_{o1} + A_2^* K_{o2}) m_{A_2} - \Gamma K_{o1} K_{o2})\right)^2 + 4\Gamma (K_{r2} m_{A^+} A^{++} + K_{o1} (2\Gamma K_{r2} + m_{A^+})) m_{A_2} (K_{o2} (2 K_{r1} m_{A_2} A_2^* + (K_{r1} A^{++} + K_{o1}) m_{A^+}) - 2 K_{o1} K_{r2} m_{A_2})\right)}\right) / \left(4\Gamma K_{o1} K_{r2} m_{A_2} - 2 K_{o2} (2\Gamma K_{r1} m_{A_2} A_2^* + \Gamma (K_{r1} A^{++} + K_{o1}) m_{A^+})\right)$$

$$i_f \rightarrow \left(F m_{A^+} (4\Gamma K_{r1} K_{r2} m_{A_2} A^{++} + m_{A^+} (\Gamma K_{o1} K_{o2} + ((K_{r1} + K_{r2}) A^{++} + K_{o1} + A_2^* K_{o2}) m_{A_2}) + \sqrt{\left((4\Gamma K_{o1} K_{r2} m_{A_2} + m_{A^+} (((K_{r1} + K_{r2}) A^{++} + K_{o1} + A_2^* K_{o2}) m_{A_2} - \Gamma K_{o1} K_{o2})\right)^2 + 4\Gamma (K_{r2} m_{A^+} A^{++} + K_{o1} (2\Gamma K_{r2} + m_{A^+})) m_{A_2} (K_{o2} (2 K_{r1} m_{A_2} A_2^* + (K_{r1} A^{++} + K_{o1}) m_{A^+}) - 2 K_{o1} K_{r2} m_{A_2})\right)}\right) / \left(K_{o2} m_{A^+}^2 - 2 (K_{r1} + K_{r2}) m_{A_2} m_{A^+} - 4\Gamma K_{r1} K_{r2} m_{A_2}\right)$$

### 4.3.4 Faradaic impedance

Faradaic impedance

$$Z_f(s) = R_{ct} + Z_{A^+}(s) + Z_{A_2}(s) + Z_A(s) + Z_s(s)$$

$$Z_f(s) \rightarrow (s + A^+(0) (K_{r1} + K_{r2}) + K_{o1} (2 \theta_A \Gamma K_{r2} M_{A^+}(s) + \theta_s \Gamma K_{o2} M_{A_2}(s) + 1) + K_{o2} (A_2(0) + \theta_s \Gamma (s + A^+(0) K_{r1}) M_{A_2}(s)) + \Gamma M_{A^+}(s) (\theta_A s K_{r2} + K_{r1} (\theta_s s + 2 A^+(0) (\theta_A + \theta_s) K_{r2} + \theta_s K_{o2} (2 A_2(0) + \theta_s s \Gamma M_{A_2}(s)))) / (f F \Gamma (A_2(0) \theta_s K_{o2} (s + 2 A^+(0) K_{r1}) \alpha_{o2} + A^+(0) \theta_A (s + 2 A^+(0) K_{r1}) K_{r2} \alpha_{r2} + A^+(0) \theta_s K_{r1} \alpha_{r1} (s + 2 A^+(0) K_{r2} + K_{o2} (2 A_2(0) + \theta_s s \Gamma M_{A_2}(s))) + K_{o1} (2 (A_2(0) \theta_s K_{o2} \alpha_{o2} + A^+(0) \theta_A K_{r2} \alpha_{r2}) + \theta_A \alpha_{o1} (s + 2 A^+(0) K_{r2} + K_{o2} (2 A_2(0) + \theta_s s \Gamma M_{A_2}(s))))))$$

$$M_{A^+} = \frac{1}{m_{A^+}} \frac{\text{th} \sqrt{\tau_{A^+}} s}{\sqrt{\tau_{A^+}} s}, \quad M_{A_2} = \frac{1}{m_{A_2}} \frac{\text{th} \sqrt{\tau_{A_2}} s}{\sqrt{\tau_{A_2}} s}$$

Transfer resistance

$$R_{ct} \rightarrow \frac{1}{f F \Gamma (\theta_A K_{o1} \alpha_{o1} + A_2(0) \theta_s K_{o2} \alpha_{o2} + A^+(0) \theta_s K_{r1} \alpha_{r1} + A^+(0) \theta_A K_{r2} \alpha_{r2})}$$

Concentration impedances

Soluble species

$$Z_{A^+}(s) \rightarrow \Gamma (\theta_s K_{r1} + \theta_A K_{r2}) R_{ct} M_{A^+}(s)$$

$Z_{A_2}(s) \rightarrow$

$$-(\theta_s \Gamma K_{o2} R_{ct} (-A^+(0) \theta_s K_{r1} \alpha_{r1} (A_2(0) K_{o2} + K_{r2} (A^+(0) - \theta_A s \Gamma M_{A^+}(s))) + \theta_A K_{o1} \alpha_{o1} (K_{r2} (\theta_A s \Gamma M_{A^+}(s) - A^+(0)) - A_2(0) K_{o2}) - A_2(0) \theta_s K_{o2} \alpha_{o2} (s + K_{o1} + K_{r1} (A^+(0) + \theta_s s \Gamma M_{A^+}(s))) - A^+(0) \theta_A K_{r2} \alpha_{r2} (s + K_{o1} + K_{r1} (A^+(0) + \theta_s s \Gamma M_{A^+}(s)))) M_{A_2}(s) / (A_2(0) \theta_s K_{o2} (s + 2 A^+(0) K_{r1}) \alpha_{o2} + A^+(0) \theta_A (s + 2 A^+(0) K_{r1}) K_{r2} \alpha_{r2} + A^+(0) \theta_s K_{r1} \alpha_{r1} (s + 2 A^+(0) K_{r2} + K_{o2} (2 A_2(0) + \theta_s s \Gamma M_{A_2}(s))) + K_{o1} (2 (A_2(0) \theta_s K_{o2} \alpha_{o2} + A^+(0) \theta_A K_{r2} \alpha_{r2}) + \theta_A \alpha_{o1} (s + 2 A^+(0) K_{r2} + K_{o2} (2 A_2(0) + \theta_s s \Gamma M_{A_2}(s))))))$$

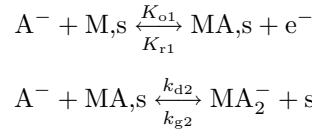
Adsorbed species

$$\begin{aligned}
Z_A(s) \rightarrow & ((K_{o1} - A^+(0) K_{r2}) R_{ct} (-A_2(0) \theta_s K_{o2} \alpha_{o2} (2 \theta_s \Gamma K_{r1} M_{A^+}(s) + 1) - A^+(0) \theta_A K_{r2} \alpha_{r2} (2 \theta_s \Gamma K_{r1} M_{A^+}(s) + 1) + \theta_A K_{o1} \alpha_{o1} \\
& (2 \theta_A \Gamma K_{r2} M_{A^+}(s) + \theta_s \Gamma K_{o2} M_{A_2}(s) + 1) + A^+(0) \theta_s K_{r1} \alpha_{r1} (2 \theta_A \Gamma K_{r2} M_{A^+}(s) + \theta_s \Gamma K_{o2} M_{A_2}(s) + 1))) / \\
& (A_2(0) \theta_s K_{o2} (s + 2 A^+(0) K_{r1}) \alpha_{o2} + A^+(0) \theta_A (s + 2 A^+(0) K_{r1}) K_{r2} \alpha_{r2} + \\
& A^+(0) \theta_s K_{r1} \alpha_{r1} (s + 2 A^+(0) K_{r2} + K_{o2} (2 A_2(0) + \theta_s s \Gamma M_{A_2}(s))) + \\
& K_{o1} (2 (A_2(0) \theta_s K_{o2} \alpha_{o2} + A^+(0) \theta_A K_{r2} \alpha_{r2}) + \theta_A \alpha_{o1} (s + 2 A^+(0) K_{r2} + K_{o2} (2 A_2(0) + \theta_s s \Gamma M_{A_2}(s))))))
\end{aligned}$$

$$\begin{aligned}
Z_s(s) \rightarrow & ((A_2(0) K_{o2} - A^+(0) K_{r1}) R_{ct} (A_2(0) \theta_s K_{o2} \alpha_{o2} (2 \theta_s \Gamma K_{r1} M_{A^+}(s) + 1) + A^+(0) \theta_A K_{r2} \alpha_{r2} (2 \theta_s \Gamma K_{r1} M_{A^+}(s) + 1) - \theta_A K_{o1} \\
& \alpha_{o1} (2 \theta_A \Gamma K_{r2} M_{A^+}(s) + \theta_s \Gamma K_{o2} M_{A_2}(s) + 1) - A^+(0) \theta_s K_{r1} \alpha_{r1} (2 \theta_A \Gamma K_{r2} M_{A^+}(s) + \theta_s \Gamma K_{o2} M_{A_2}(s) + 1))) / \\
& (A_2(0) \theta_s K_{o2} (s + 2 A^+(0) K_{r1}) \alpha_{o2} + A^+(0) \theta_A (s + 2 A^+(0) K_{r1}) K_{r2} \alpha_{r2} + \\
& A^+(0) \theta_s K_{r1} \alpha_{r1} (s + 2 A^+(0) K_{r2} + K_{o2} (2 A_2(0) + \theta_s s \Gamma M_{A_2}(s))) + \\
& K_{o1} (2 (A_2(0) \theta_s K_{o2} \alpha_{o2} + A^+(0) \theta_A K_{r2} \alpha_{r2}) + \theta_A \alpha_{o1} (s + 2 A^+(0) K_{r2} + K_{o2} (2 A_2(0) + \theta_s s \Gamma M_{A_2}(s))))))
\end{aligned}$$

## 4.4 Copper dissolution in HCl

### 4.4.1 Mechanism [5, 1, 2]



### 4.4.2 Kinetic equations

Langmuir isotherm:  $K_{o1} = k_{o1} \exp(\alpha_{o1} f E)$ ,  $K_{r1} = k_{r1} \exp(-\alpha_{r1} f E)$

#### Transformation rates

$$v_{A^-}(t) = -v_1(t) - v_2(t), v_{MA_2^-}(t) = v_2(t), v_s(t) = -v_1(t) + v_2(t), v_{MA}(t) = v_1(t) - v_2(t)$$

#### Mass balance equations

Flux of soluble species

$$J_{A^-}(0, t) = v_{A^-}(t), J_{MA_2^-}(0, t) = v_{MA_2^-}(t)$$

Rate of production of adsorbed species

$$\frac{d\theta_s(t)}{dt} = \frac{v_s(t)}{\Gamma}, \frac{d\theta_{MA}(t)}{dt} = \frac{v_M(t)}{\Gamma}$$

#### Current density vs. step rates

$$i_f(t) = F v_1(t)$$

#### Step rates

$$\begin{aligned}
v_1(t) &= A^-(0, t) \theta_s(t) \Gamma K_{o1}(t) - \theta_{MA}(t) \Gamma K_{r1}(t) \\
v_2(t) &= A^-(0, t) \theta_{MA}(t) \Gamma k_{d2} - MA_2^-(0, t) \theta_s(t) \Gamma k_{g2}
\end{aligned}$$

### 4.4.3 Steady-state conditions

#### Steady-state equations

Soluble species

$$J_{A^-}(0) = - \left( A^{-*} - A^-(0) \right) m_{A^-}, J_{MA_2^-}(0) = - \left( MA_2^{-*} - MA_2^-(0) \right) m_{MA_2^-}$$

Adsorbed species

$$d\theta_s/dt = 0, \theta_{MA} + \theta_s = 1$$

#### Steady-state solutions

Soluble species

$$A^-(0) \rightarrow (m_{A^-} ((K_{r1} - A^{-*} (k_{d2} + K_{o1})) m_{MA_2^-} + k_{g2} (m_{A^-} A^{-*} + \Gamma K_{r1} + MA_2^{-*} m_{MA_2^-})) - \sqrt{((\Gamma k_{g2} K_{r1} m_{A^-} - (4 \Gamma k_{d2} K_{r1} + ((k_{d2} + K_{o1}) A^{-*} + MA_2^{-*} k_{g2} + K_{r1}) m_{A^-}) m_{MA_2^-})^2 + 4 \Gamma (2 \Gamma k_{d2} K_{r1} + (k_{d2} A^{-*} + K_{r1}) m_{A^-}) m_{MA_2^-} (k_{g2} (2 K_{o1} m_{MA_2^-} MA_2^{-*} + (K_{o1} A^{-*} + K_{r1}) m_{A^-}) - 2 k_{d2} K_{r1} m_{MA_2^-}))}) / (k_{g2} m_{A^-}^2 - 2 (2 \Gamma k_{d2} K_{o1} + (k_{d2} + K_{o1}) m_{A^-}) m_{MA_2^-})$$

$$MA_2^-(0) \rightarrow (k_{g2} (\Gamma K_{r1} - MA_2^{-*} m_{MA_2^-}) m_{A^-}^2 - \sqrt{((\Gamma k_{g2} K_{r1} m_{A^-} - (4 \Gamma k_{d2} K_{r1} + ((k_{d2} + K_{o1}) A^{-*} + MA_2^{-*} k_{g2} + K_{r1}) m_{A^-}) m_{MA_2^-})^2 + 4 \Gamma (2 \Gamma k_{d2} K_{r1} + (k_{d2} A^{-*} + K_{r1}) m_{A^-}) m_{MA_2^-} (k_{g2} (2 K_{o1} m_{MA_2^-} MA_2^{-*} + (K_{o1} A^{-*} + K_{r1}) m_{A^-}) - 2 k_{d2} K_{r1} m_{MA_2^-}))}) m_{A^-} + m_{MA_2^-} (4 (2 \Gamma k_{d2} K_{o1} + (k_{d2} + K_{o1}) m_{A^-}) m_{MA_2^-} MA_2^{-*} + m_{A^-} (4 \Gamma k_{d2} K_{o1} A^{-*} + ((k_{d2} + K_{o1}) A^{-*} + K_{r1}) m_{A^-}))) / (4 (2 \Gamma k_{d2} K_{o1} + (k_{d2} + K_{o1}) m_{A^-}) m_{MA_2^-}^2 - 2 k_{g2} m_{A^-}^2 m_{MA_2^-})$$

Adsorbed species

$$\theta_{MA} \rightarrow (-\Gamma k_{g2} (2 K_{o1} A^{-*} + K_{r1}) m_{A^-} - (k_{g2} (4 \Gamma K_{o1} + m_{A^-}) MA_2^{-*} + ((k_{d2} + K_{o1}) A^{-*} + K_{r1}) m_{A^-}) m_{MA_2^-} + \sqrt{((\Gamma k_{g2} K_{r1} m_{A^-} - (4 \Gamma k_{d2} K_{r1} + ((k_{d2} + K_{o1}) A^{-*} + MA_2^{-*} k_{g2} + K_{r1}) m_{A^-}) m_{MA_2^-})^2 + 4 \Gamma (2 \Gamma k_{d2} K_{r1} + (k_{d2} A^{-*} + K_{r1}) m_{A^-}) m_{MA_2^-} (k_{g2} (2 K_{o1} m_{MA_2^-} MA_2^{-*} + (K_{o1} A^{-*} + K_{r1}) m_{A^-}) - 2 k_{d2} K_{r1} m_{MA_2^-}))}) / (4 \Gamma k_{d2} K_{r1} m_{MA_2^-} - 2 k_{g2} (2 \Gamma K_{o1} m_{MA_2^-} MA_2^{-*} + \Gamma (K_{o1} A^{-*} + K_{r1}) m_{A^-}))$$

Current density

$$i_f \rightarrow -(F m_{A^-} (\Gamma k_{g2} K_{r1} m_{A^-} + (4 \Gamma k_{d2} K_{o1} A^{-*} + ((k_{d2} + K_{o1}) A^{-*} + MA_2^{-*} k_{g2} + K_{r1}) m_{A^-}) m_{MA_2^-} - \sqrt{((\Gamma k_{g2} K_{r1} m_{A^-} - (4 \Gamma k_{d2} K_{r1} + ((k_{d2} + K_{o1}) A^{-*} + MA_2^{-*} k_{g2} + K_{r1}) m_{A^-}) m_{MA_2^-})^2 + 4 \Gamma (2 \Gamma k_{d2} K_{r1} + (k_{d2} A^{-*} + K_{r1}) m_{A^-}) m_{MA_2^-} (k_{g2} (2 K_{o1} m_{MA_2^-} MA_2^{-*} + (K_{o1} A^{-*} + K_{r1}) m_{A^-}) - 2 k_{d2} K_{r1} m_{MA_2^-}))}) / (2 k_{g2} m_{A^-}^2 - 4 (2 \Gamma k_{d2} K_{o1} + (k_{d2} + K_{o1}) m_{A^-}) m_{MA_2^-})$$

### 4.4.4 Faradaic impedance

Faradaic impedance

$$Z_f(s) = R_{ct} + Z_{A^-}(s) + Z_{MA}(s) + Z_s(s)$$



$$Z_f(s) \rightarrow (s + MA_2^-(0) k_2 + K_1 + K_1 (A^-(0) + \theta_s \Gamma (s + 2 MA_2^-(0) k_2) M_{A^-}(s)) + k_2 (A^-(0) + \Gamma (2 A^-(0) (\theta_{MA} + \theta_s) K_1 + \theta_{MA} (s + 2 K_1)) M_{A^-}(s)) + \theta_s \Gamma k_2 (s + K_1 + K_1 (A^-(0) + \theta_s s \Gamma M_{A^-}(s))) M_{MA_2^-}(s)) / (f F \Gamma (\theta_{MA} K_1 \alpha_1 + A^-(0) \theta_s K_1 \alpha_1) (s + k_2 (A^-(0) + \theta_{MA} s \Gamma M_{A^-}(s)) + k_2 (MA_2^-(0) + \theta_s s \Gamma M_{MA_2^-}(s))))$$

$$M_{A^-} = \frac{1}{m_{A^-}} \frac{\text{th} \sqrt{\tau_{A^-} s}}{\sqrt{\tau_{A^-} s}}, \quad M_{MA_2^-} = \frac{1}{m_{MA_2^-}} \frac{\text{th} \sqrt{\tau_{MA_2^-} s}}{\sqrt{\tau_{MA_2^-} s}}$$

Charge transfer resistance

$$R_{ct} = \frac{1}{f F \Gamma (A^-(0) \theta_s K_{o1} \alpha_{o1} + \theta_{MA} K_{r1} \alpha_{r1})}$$

Concentration impedances

Soluble species

$$Z_{A^-}(s) = \frac{\theta_s \Gamma K_{o1} R_{ct} M_{A^-}(s) (s + 2 A^-(0) k_{d2} + k_{g2} (2 MA_2^-(0) + \theta_s s \Gamma M_{MA_2^-}(s)))}{s + k_{d2} (A^-(0) + \theta_{MA} s \Gamma M_{A^-}(s)) + k_{g2} (MA_2^-(0) + \theta_s s \Gamma M_{MA_2^-}(s))}$$

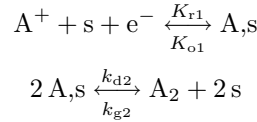
Adsorbed species

$$Z_{MA}(s) = \frac{K_{r1} R_{ct} (1 + 2 \theta_{MA} \Gamma k_{d2} M_{A^-}(s) + \theta_s \Gamma k_{g2} M_{MA_2^-}(s))}{s + k_{d2} (A^-(0) + \theta_{MA} s \Gamma M_{A^-}(s)) + k_{g2} (MA_2^-(0) + \theta_s s \Gamma M_{MA_2^-}(s))}$$

$$Z_s(s) = \frac{A^-(0) K_{o1} R_{ct} (1 + 2 \theta_{MA} \Gamma k_{d2} M_{A^-}(s) + \theta_s \Gamma k_{g2} M_{MA_2^-}(s))}{s + k_{d2} (A^-(0) + \theta_{MA} s \Gamma M_{A^-}(s)) + k_{g2} (MA_2^-(0) + \theta_s s \Gamma M_{MA_2^-}(s))}$$

## 4.5 (V-T) reaction with mass transfer limitation

### 4.5.1 Mechanism



### 4.5.2 Kinetic equations

Langmuir isotherm:  $K_{r1} = k_{r1} \exp(-\alpha_{r1} f E)$ ,  $K_{o1} = k_{o1} \exp(\alpha_{o1} f E)$

Transformation rates

$$v_{A^+}(t) = -v_1(t), v_{A_2}(t) = v_2(t), v_s(t) = -v_1(t) + 2 v_2(t), v_A(t) = v_1(t) - 2 v_2(t)$$

Mass balance equations

Flux of soluble species

$$J_{A^+}(0, t) = v_{A^+}(t), J_{A_2}(0, t) = v_{A_2}(t)$$

Rate of production of adsorbed species,

$$\frac{d\theta_s(t)}{dt} = \frac{v_s(t)}{\Gamma}, \quad \frac{d\theta_A(t)}{dt} = \frac{v_A(t)}{\Gamma}$$

## Current density vs. step rates

$$i_f(t) = -F v_1(t)$$

## Step rates

$$v_1(t) = -\theta_A(t) \Gamma K_{o1}(t) + A^+(0, t) \theta_s(t) \Gamma K_{r1}(t), \quad v_2(t) = \theta_A(t)^2 \Gamma^2 k_{d2} - A_2(0, t) \theta_s(t)^2 \Gamma^2 k_{g2}$$

## 4.5.3 Steady-state conditions

### Steady-state equations

Soluble species

$$J_{A^+}(0) = -\left(A^{+*} - A^+(0)\right) m_{A^+}, \quad J_{A_2}(0) = (A_2^* - A_2(0)) m_{A_2}$$

Adsorbed species

$$d\theta_A/dt = 0, \quad \theta_A + \theta_s = 1$$

### Steady-state solutions

Soluble species:  $A^+(0)$  and  $A_2(0)$  are solutions of cubic equations.

$A^+(0)$ :

$$2 k_{d2} \left( A^+(0) \Gamma K_{r1} + \left( -A^{+*} + A^+(0) \right) m_{A^+} \right)^2 m_{A_2} = \left( A^{+*} - A^+(0) \right) \left( K_{o1} + A^+(0) K_{r1} \right)^2 m_{A^+} m_{A_2} + k_{g2} \left( \Gamma K_{o1} + \left( A^{+*} - A^+(0) \right) m_{A^+} \right)^2 \left( \left( A^{+*} - A^+(0) \right) m_{A^+} + 2 A_2^* m_{A_2} \right)$$

$A_2(0)$ :

$$4 (A_2^* - A_2(0))^2 K_{r1} m_{A_2}^2 \left( \left( K_{o1} + A^{+*} K_{r1} \right) m_{A^+} + (A_2^* - A_2(0)) K_{r1} m_{A_2} \right) + k_{d2} \left( 2 (A_2^* - A_2(0)) m_{A^+} m_{A_2} + \Gamma K_{r1} \left( A^{+*} m_{A^+} + 2 (A_2^* - A_2(0)) m_{A_2} \right) \right)^2 = m_{A^+}^2 \left( - \left( (A_2^* - A_2(0)) \left( K_{o1} + A^{+*} K_{r1} \right)^2 m_{A_2} \right) + A_2(0) k_{g2} \left( \Gamma K_{o1} + 2 (-A_2^* + A_2(0)) m_{A_2} \right)^2 \right)$$

Adsorbed species:  $\theta_A$  is solution of a cubic equation

$$2 A_2^* \Gamma k_{g2} m_{A_2} (-m_{A^+} + \Gamma K_{r1} (-1 + \theta_A)) (-1 + \theta_A)^2 + m_{A^+} \left( m_{A_2} + \Gamma^2 k_{g2} (-1 + \theta_A)^2 \right) \left( A^{+*} K_{r1} (-1 + \theta_A) + K_{o1} \theta_A \right) + 2 \Gamma k_{d2} m_{A_2} \theta_A^2 (m_{A^+} + K_{r1} (\Gamma - \Gamma \theta_A)) = 0$$

Current density:  $i_f$  is solution of a cubic equation

$$2 F i_f (2 \Gamma k_{d2} + K_{o1}) K_{r1} m_{A^+} \left( i_f + A^{+*} F m_{A^+} \right) m_{A_2} + \left( i_f + 2 F \Gamma^2 k_{d2} \right) K_{r1}^2 \left( i_f + A^{+*} F m_{A^+} \right)^2 m_{A_2} + m_{A^+}^2 \left( F i_f (2 i_f k_{d2} + F K_{o1}^2) m_{A_2} + k_{g2} (i_f - F \Gamma K_{o1})^2 (i_f - 2 A_2^* F m_{A_2}) \right) = 0$$

#### 4.5.4 Faradaic impedance

Faradaic impedance

$$Z_f(s) = R_{ct} + Z_{A^+}(s) + Z_A(s) + Z_s(s)$$

$$Z_f(s) \rightarrow (s + K_{o1} + K_{r1} (A^+(0) + \theta_s s \Gamma M_{A^+}(s)) + 4 \theta_A \Gamma k_{d2} (\theta_s \Gamma K_{r1} M_{A^+}(s) + 1) + \theta_s \Gamma k_{g2} (4 (\theta_s \Gamma K_{r1} M_{A^+}(s) A_2(0) + A_2(0)) + \theta_s \Gamma (s + K_{o1} + K_{r1} (A^+(0) + \theta_s s \Gamma M_{A^+}(s))) M_{A_2}(s))) / (f F \Gamma (\theta_A K_{o1} \alpha_{o1} + A^+(0) \theta_s K_{r1} \alpha_{r1}) (s + 4 \theta_A \Gamma k_{d2} + \theta_s \Gamma k_{g2} (4 A_2(0) + \theta_s s \Gamma M_{A_2}(s))))$$

$$M_{A^+} = \frac{1}{m_{A^+}} \frac{\text{th} \sqrt{\tau_{A^+} s}}{\sqrt{\tau_{A^+} s}}, \quad M_{A_2} = \frac{1}{m_{A_2}} \frac{\text{th} \sqrt{\tau_{A_2} s}}{\sqrt{\tau_{A_2} s}}$$

Charge transfer resistance

$$R_{ct} = \frac{1}{f F \Gamma (\theta_A K_{o1} \alpha_{o1} + A^+(0) \theta_s K_{r1} \alpha_{r1})}$$

Concentration impedances Soluble species

$$Z_{A^+}(s) = \theta_s \Gamma K_{r1} R_{ct} M_{A^+}(s)$$

Adsorbed species

$$Z_A(s) = \frac{K_{o1} R_{ct} (1 + \theta_s^2 \Gamma^2 k_{g2} M_{A_2}(s))}{s + 4 \theta_A \Gamma k_{d2} + \theta_s \Gamma k_{g2} (4 A_2(0) + \theta_s s \Gamma M_{A_2}(s))}$$

$$Z_s(s) = \frac{A^+(0) K_{r1} R_{ct} (1 + \theta_s^2 \Gamma^2 k_{g2} M_{A_2}(s))}{s + 4 \theta_A \Gamma k_{d2} + \theta_s \Gamma k_{g2} (4 A_2(0) + \theta_s s \Gamma M_{A_2}(s))}$$



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